

PII: S0040-4039(97)01521-9

## **Approach Towards an EPC-Synthesis of Nodusmicln - IV. 1**

## **Construction of the Highly Hindered Trisubstituted Double Bond of Nodusmicin. 2**

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**Abstract:** The highly substituted subunit 6 of our convergent synthesis of nodusmicin was prepared utilizing Evans' oxazolidinone protocol. Compound 6 was then adjoined to the decalin subunit Z using Martin's olefination method. To improve on the formation of the hindered trisubslituted double bond, the following model reactions were devised: reductive coupling of ester and keto functionality of tricycle 16 led to the α-hydroxyketone, which was converted to the cyclic sulfate 19 via the diol. Sodium naphthalide then converted 19 to the olefins **20.**  © 1997 Elsevier Science Ltd.

The increasing resistance of bacterial pathogens against antibiotics necessitates the development of new and effective antibiotic species.<sup>3</sup> We therefore chose nodusmicin  $1^{4.5}$  due to its antibacterial activity and unusual structure as target of a convergent enantiomerically pure synthesis.

After completing the synthesis of the enantiomerically pure *cis-decalin* part of nodusmicin, 1 we prepared the second subunit, an enantiomerically pure, highly substituted, five-carbon chain using Evans' enantioselective aldol protocol.<sup>6</sup> We report this synthesis and our efforts to connect the subunits with the formation of a highly congested trisubstituted double bond.



The boron enolate of oxopropyloxazolidinone<sup>6</sup> was added to  $(R)$ -2-benzyloxypropanal<sup>7</sup> yielding the oxazolidinone derivative  $2^2$  X-ray analysis of the recrystallized compound 2 revealed that the expected stereoisomer was produced.<sup>8</sup> After transforming the alcohol 2 into its tert-butyldimethylsilylether reduction by LiBH $_4^9$  afforded the primary alcohol 3.

In a preliminary attempt, alcohol  $3$  was converted into the p-tolylsulfone derivative  $5$  via the iodide 4 according to Corey et al.<sup>10</sup> and consecutive treatment with sodium p-tolylsulfinate.<sup>11</sup> Several attempts to achieve the intended Julia reaction<sup>12</sup> with enantiomerically pure ketone  $\mathbb{Z}^1$  failed due to severe steric hindrance at the carbonyl position. We then turned our attention to Martin's *recently*  published olefination method<sup>13</sup> to prepare the desired trisubstituted double bond.



Thus, enantiomerically pure Z was converted to the vinyl lithium compound via the sulfonylhydrazone  $\mathbf{g}^{14}$  and reacted with aldehyde  $\mathbf{g}$ , prepared by Dess-Martin oxidation<sup>15</sup> of alcohol 3. A 1:1 mixture of allylalcohols 9 was obtained in 46% yield. This mixture was converted to the corresponding xanthates, and thermal [3,3]-sigmatropic rearrangement led to the dithiocarbonates in excellent yields. This rearrangement was highly stereoselective as each of the chromatographically separated xanthates led exclusively to a single dithiocarbonate. Subsequent desulfuration by tributylstannane afforded the olefins 10 : 11 as a 5 : 1 (E : Z ) mixture regardless of the configuration of the double bond in the starting dithiocarbonate. The configuration of the double bond was deduced by NMR spectroscopy.<sup>16</sup> To connect the third and last subunit of our convergent synthesis, the olefins were transformed as follows. Partial desilylation with fluoride at 0°C cleaved the cyclic ketal. At this point, the two bicyclic olefins were separated by chromatography. The secondary alcohol of the purified E-olefin was protected as methoxymethyl ether 12. Treatment with tetrabutylammonium fluoride at elevated temperature cleaved the exocyclic silyl ether, and the liberated alcohol was esterified with  $\alpha$ -methoxy acrylic acid.<sup>17</sup> Thus all the subunits of our projected synthesis were adjoined.

Although we have succeeded in connecting the subunits in our projected synthesis, the modera-

te yields of the decisive coupling of the first two subunits tempted us to develop a new method to prepare the congested trisubstituted double bond by means of model compounds. Therefore, we chose 5-acetyl-7-benzyloxy-10-hydroxybicyclo[4.4.0]decan-3-one 14<sup>18</sup> as the simplified model of the decalin moiety. As the smallest relevant open chained subunit the isobutyl group was elected.



To diminish the free enthalpy of activation, we turned our attention to intramolecular carbonyl coupling reactions. Thus, diketone 14 was converted to the cyclic ketal 15 and debenzylated by hydrogenolysis. Esterification with isobutyric acid yielded 16. Unfortunately, several attempts at the McMurry reaction<sup>19</sup> were tried to no avail. Therefore, we examined other variants of the pinacol type reaction.<sup>20,21</sup> By adjusting the reaction conditions to -78°C and 1,05 equivalents of lithium naphthalide,<sup>22</sup> the hydroxyketone 17 was obtained in good yields. After protection of the secondary alcohol of 17, reduction with lithium aluminum hydride led to the vicinal diol 18, which was converted to the cyclic sulfate 19 according to literature.<sup>23</sup> This cyclic sulfate was treated with sodium naphthalide affording a mixture of E- and Z-olefin  $20$  resp.(2,4 : 1) in high yield,<sup>24</sup> which was transformed to the bicyclic ketones 21. We are now adjusting this method to the fully substituted subunits.

Acknowledgement : This work was financially supported by the Fonds zur Förderung der wissenschaftlichen Forschung (project P 8872-CHE). M.G. thanks the University of Vienna for a scholarship for the year 1996.

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- 21 a) With Li/NH<sub>3</sub> at -78°C, the keto group of 16 was reduced (40%) and only small amounts of the desired 18 (10%) were isolated. Compound 17 was not detected, b) Under acyloin reaction conditions (Na/K, TMSCI, ether, rt) intramolecular coupling of 16 proceeded in good yields but led, due to overreduction, to the silylenolethar, c) With excess lithium naphthalide in THF at -78 $\degree$ C, 16 was converted via hydroxyketone 17 by deoxygenation to the corresponding ketone.
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- 24 Beels, C.MD.; Coleman, M.J.; Taylor, R.J.K. *Synlett* 1990, 479-480; very recently conversion of cyclic sulfates to olefins via lithium iodide has been reported: Jang, O.J.; Joo, Y.H. Synlett 1997, 279-280. Significant NMR data of E-olefin 21 <sup>1</sup>H NMR (CDCI<sub>3</sub>):  $\delta = 1.61$  (d, 3H, J<sub>1'.3</sub>.= 1,3 Hz, H-1'), 4.94 (d (br), 1H, J<sub>3',4</sub>.= 9 Hz, H-3'); <sup>13</sup>C NMR (CDCI<sub>3</sub>):  $\delta = 14.4$  C-1', 134 C-3', 134.7 C-2' and of Z-olefin 21. <sup>1</sup>H NMR (CDCI<sub>3</sub>):  $\delta = 1.67$  (d, 3H, J<sub>1'.3</sub> = 1.3 Hz, H-1 ), 4.93 (d (br), 1H,  $J_{3',4}$  = 9 Hz, H-3'); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta = 18.8$  C-1', 133.9 C-2', 134.7 C-3'.